

Modeling studies of NO- γ emissions of sprites

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[1] Motivated by observations of NO- γ emissions: $NO(A^2\Sigma^+) \rightarrow NO(X^2\Pi_r) + h\nu$ from streamer discharges in laboratory experiments, we develop a model to study these emissions associated with streamer discharges in air at different pressures. The modeling results indicate that the $NO(A^2\Sigma^+)$ species in sprite streamers at 70 km altitude are mostly produced by interaction of $N_2(A^3\Sigma_u^+)$ metastable species with high-density ambient NO($X^2\Pi_r$) molecules. Analysis of the production and loss mechanisms for the upper excited states leading to NO- γ and N₂ LBH emissions demonstrates that the total intensity of NO- γ emissions associated with sprites is substantially weaker than that of the N₂ LBH emissions. The same conclusion can be drawn for elves following similar analysis. It is demonstrated that the presented modeling approach and analysis are consistent with prior knowledge on NO- γ emissions from aurora, indicating that to observe the NO- γ emissions from aurora, the ambient NO density must be highly enhanced in the auroral region. Citation: Liu, N., and V. P. Pasko (2007), Modeling studies of NO- γ emissions of sprites, Geophys. Res. Lett., 34, L16103, doi:10.1029/2007GL030352.

1. Introduction

[2] Sprites are spectacular luminous discharges, which appear in the altitude range of ~40 to 90 km above thunderstorms [*Sentman et al.*, 1995]. Optical measurements of sprites indicate that four emission band systems of N₂ and N₂⁺ appear in the sprite spectrum. They are red emissions from the first positive band system of N₂ (1PN₂) [e.g., *Mende et al.*, 1995; *Hampton et al.*, 1996], blue emissions of second positive band system of N₂ (2PN₂) and the first negative band system of N₂⁺ (1NN₂⁺) [e.g., *Armstrong et al.*, 1998; *Suszcynsky et al.*, 1998], and far-UV emissions of LBH band system of N₂ [e.g., *Mende et al.*, 2006].

[3] Telescopic imaging and high-speed video observations reveal that decameter-scale thin structures are present in sprites [e.g., *Stanley et al.*, 1999; *Gerken et al.*, 2000; *Cummer et al.*, 2006; *McHarg et al.*, 2007; *Stenbaek-Nielsen et al.*, 2007], which are in a good agreement with the streamer mechanism of sprites [*Pasko*, 2007, and references therein].

[4] Spectroscopic studies of sprites provide insightful information about the sprite discharges. After analyzing simultaneous recordings of various emission band systems of sprites by the ISUAL instrument on the FORMOSAT-2 satellite, *Kuo et al.* [2005] conclude that the strength of the

driving electric field, and the mean energy of electrons in sprites are \sim 3 times of the conventional breakdown field of air, and 6.2–9.2 eV, respectively, consistent with streamer modeling results. These results are further supported by the direct comparison of intensity ratios of different emissions in the ISUAL spectrophotometric data with modeling results on sprite streamer emissions [*Liu et al.*, 2006].

[5] The study presented in this paper is motivated by spectroscopic work on streamer discharges in laboratory experiments [e.g., *Tochikubo and Teich*, 2000; *Ono and Oda*, 2005]. In laboratory experiments N₂ LBH emissions are generally obscured by the strong emissions of NO- γ system: NO($A^2\Sigma^+$) \rightarrow NO($X^2\Pi_r$) + $h\nu$ [e.g., *Teich*, 1993; *Tochikubo and Teich*, 2000], the wavelengths of which are in the range of 200–290 nm. The purpose of this paper is to discuss NO chemistry in low pressure streamer discharges as applicable to sprite phenomenon and to report quantitative modeling results on NO- γ emissions from sprite streamers.

2. Model Formulation

[6] The streamer model equations used in this study are the same as in the work by *Liu and Pasko* [2004], and the simulation domain is identical to the one of *Liu and Pasko* [2005, Figure 1b]. A chemical model to study NO chemistry in streamers is solved simultaneously with the streamer model. This model includes 10 neutral species: N₂, O₂, N(²D), N(⁴S), O(³P), N₂(A³\Sigma⁺_u), N₂(B³\Pi_g), N₂(C³\Pi_u), NO(X²Π_r), and NO(A²Σ⁺). Important reactions for NO chemistry and NO- γ emissions are listed in Table 1.

[7] The direct excitation of NO($A^2\Sigma^+$) by electron impact is negligible, and these species are excited mainly through resonant energy transfer: N₂($A^3\Sigma_u^+$) + NO($X^2\Pi_r$) \rightarrow NO($A^2\Sigma^+$) + N₂($X^1\Sigma_g^+$), where N₂($A^3\Sigma_u^+$) is a metastable state [*Simek*, 2002, and references therein]. N₂($A^3\Sigma_u^+$) has a relatively low excitation energy threshold ~6 eV and a lifetime of 2 s [*Simek*, 2003]. An important role is played by N₂($A^3\Sigma_u^+$) in the streamer discharges, especially in the postdischarge period, including population of the upper excited states leading to 1PN₂ and 2PN₂ emissions by energy pooling reactions, and by generating free electrons through reaction with N₂ singlet metastables [e.g., *Guerra et al.*, 2001; *Simek*, 2003]. The NO- γ emission is frequently used to monitor the evolution of N₂($A^3\Sigma_u^+$) species [*Simek et al.*, 1998].

[8] For streamer discharges in N₂/O₂ mixtures, NO is produced by the reaction: N(²D) + O₂ \rightarrow NO($X^{2}\Pi_{r}$) + O(³P). The species N(²D) are generated by electron impact dissociation of N₂ molecules. There is a lack of studies on dissociation cross sections of N₂ by electron impact in the energy range from the threshold (~10 eV) to several tens eV. We assume that N(²D) and N(⁴S) atoms are generated with

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Reaction Process or Index	Reaction	Rate Coefficient and References ^a
	Electron Collision Reacti	ons
1	$e + N_2 \rightarrow e + N(^4S) + N(^2D)$	f(E/N) [Benilov and Naidis, 2003; Zipf and McLaughlin, 1978]
2	$e + O_2 \rightarrow e + O(^3P) + O(^3P)$	f(E/N) [Benilov and Naidis, 2003]
3	$e + N_2 \rightarrow e + N_2(A^3 \Sigma_u^+)$	f(E/N) [Aleksandrov et al., 1995]
4	$e + N_2 \rightarrow e + N_2(B^3 \Pi_a)$	f(E/N) [Moss et al., 2006]
5	$e + N_2 \rightarrow e + N_2(C^3 \Pi_u)$	f(E/N) [Moss et al., 2006]
6	$e + \operatorname{NO}(X^2 \Pi_r) \to e + \operatorname{NO}(A^2 \Sigma^+)$	f(E/N) [Simek, 2002]
	Chemical Reactions	
7	$N(^{2}D) + O_{2} \rightarrow NO(X^{2}\Pi_{*}) + O(^{3}P)$	$5.20 \times 10^{-18} \text{ m}^3 \text{s}^{-1}$ [Zhao et al., 2005]
8	$N_2(A^3\Sigma_{*}^+) + O(^3P) \rightarrow NO(X^2\Pi_{*}) + N(^2D)$	$7 \times 10^{-18} \text{ m}^3 \text{s}^{-1}$ [Guerra et al., 2001]
9	$N(^{2}D) + NO \rightarrow N_{2} + O(^{3}P)$	$6.0 \times 10^{-17} \text{ m}^3 \text{s}^{-1}$ [Zhao et al., 2005]
	Excitation	
10	$\mathrm{N}_2(A^3\Sigma_u^+) + \mathrm{NO}(X^2\Pi_r) \to \mathrm{NO}(A^2\Sigma^+) + \mathrm{N}_2(X^1\Sigma_g^+)$	$8.75 \times 10^{-17} \text{ m}^3 \text{s}^{-1}$ [Simek, 2003]
	Ouenching	
11	$N(^{2}D) + N_{2} \rightarrow N(^{4}S) + N_{2}$	$1.70 \times 10^{-20} \text{ m}^3 \text{s}^{-1}$ [Zhao et al., 2005]
12	$N_2(A^3\Sigma_{\mu}^+) + O_2 \rightarrow N_2 + O_2$	$8.75 \times 10^{-19} \text{ m}^3 \text{s}^{-1}$ [Guerra et al., 2001]
13	$N_2(A^3\Sigma_u^+) + O_2 \rightarrow N_2 + 2O(^3P)$	$1.63 \times 10^{-18} \text{ m}^3 \text{s}^{-1}$ [Guerra et al., 2001]
14	$NO(A^2\Sigma^+) + O_2 \rightarrow NO(X^2\Pi_r) + O_2$	$1.62 \times 10^{-16} \text{ m}^3 \text{s}^{-1}$ [Simek, 2003]
	Radiative Transition	
15	$N_2(B^3\Pi_a) \rightarrow N_2(A^3\Sigma_{\mu}^+) + h\nu$	$1.7 \times 10^5 \text{ s}^{-1}$ [Liu and Pasko, 2004]
16	$\tilde{N_2(C^3\Pi_u)} \rightarrow \tilde{N_2(B^3\Pi_g)} + h\nu$	$2.0 \times 10^{7} \text{s}^{-1}$ [Liu and Pasko, 2004]
17	$\tilde{NO}(A^2 \Sigma^+) \rightarrow \tilde{NO}(X^2 \Pi_r) + h\nu$	$5 \times 10^6 \text{ s}^{-1}$ [Simek et al., 1998]

Table 1. List of Reactions Important for NO- γ Emissions

^aHere f(E/N) denotes function of reduced electric field; neutral temperature is 300 K.

an approximately equal ratio by electron collisions in this energy range [*Zipf and McLaughlin*, 1978].

[9] The natural lifetime of NO($A^2\Sigma^+$) is ~0.2 μ s, and these species are mainly quenched by O₂ with a coefficient 1.62×10^{-16} m³/s [*Simek*, 2003]. The resulting quenching altitude is ~35 km that is much lower than that of N₂($a^1\Pi_g$) state leading to LBH N₂ emissions (~77 km [*Liu and Pasko*, 2005, and references therein]).

3. Results and Discussion

[10] Before presenting modeling results on streamers developing at sprite altitudes, it is worthwhile to analyze the relative production rates of NO($A^2\Sigma^+$) and N₂($a^1\Pi_g$) excited molecules using similarity relations. It is known that the streamer timescales and electron density approximately scale with the air density (N) as $\sim N^{-1}$ and $\sim N^2$ [Liu and Pasko, 2004]. Table 2 summarizes the scaling laws for the reaction rates and product densities of several reactions leading to production of NO($A^2\Sigma^+$) shown in Table 1 and $N_2(a^1\Pi_{\alpha})$ molecules. As an example, the densities of the two reactants of reaction 1 scale with N as $\sim N^2$ and $\sim N$, respectively. The resulting reaction rate scales as $\sim N^3$, and the density of the products scales as $\sim N^2$ due to N^{-1} scaling of timescales for streamers. For the same composition of air, it is expected that the relative produced density of $NO(A^2\Sigma^+)$ (~ N^3) with respect to $N_2(a^1\Pi_{\sigma})$ (~ N^2) molecules is lower for streamers at lower pressures than those at high pressures. In addition to production, the quenching effects (discussed below) are very important for derivation of quantitative information about time dynamics of $NO(A^2\Sigma^+)$ and $N_2(a^1\Pi_g)$ densities. The above similarity analysis assumes that there is no background density of NO, and from modeling results presented below it appears that

the NO densities produced by streamer at 70 km altitude are negligible in comparison with ambient NO densities.

[11] The ambient NO is present in the upper atmosphere and its density at 70 km altitude is on the order of 10^{13} – 10^{14} m^{-3} depending on particular atmospheric conditions [e.g., Atreva, 1981; Kumar et al., 1995; Gordley et al., 1996]. One study found that the NO density can be as large as 2×10^{14} m⁻³ at 70 km altitude, or about 100 ppb [Atreya, 1981]. We use this value as the initial density of NO for our streamer simulations at 70 km altitude. Corresponding results are shown in Figure 1. The NO density stays at this value during the whole simulation, indicating that the production of NO in the streamer is unimportant in comparison with the ambient NO density. We note that NO can be destructed by O_3 produced in corona discharges in laboratory experiments. However, estimates indicate that this destruction within the duration of ~ 0.5 ms of the simulated streamer is negligible due to the large ambient density of NO at 70 km altitude. In the streamer channel, the densities of atomic oxygen and $N_2(A^3\Sigma_{\mu}^+)$ are much higher than the rest of species except ambient NO.

Table 2. Scaling of Densities of NO($A^2\Sigma^+$) and N₂($a^1\Pi_g$) With Air Density N

Reaction or Reaction Index	Reaction Rate	Scaling of Product
	NO- γ	
1	$\sim [e][N_2]$	$\sim N^2$
2	$\sim [e][O_2]$	$\sim N^2$
3	$\sim [e][N_2]$	$\sim N^2$
7	$\sim [N(^2D)][O_2]$	$\sim N^2$
8	$\sim [N_2(A^3\Sigma_u^+)][O(^3P)]$	$\sim N^3$
10	$\sim [N_2(A^3\Sigma_u^+)][NO(X^2\Pi_r)]$	$\sim N^3$
	No LBH	
$e + N_2 \rightarrow N_2(a^1 \Pi_a)$	$\sim [e][N_2]$	$\sim N^2$



Figure 1. (a) Density profiles of various species along the central axis of the model streamer at 70 km altitude and (b) intensity profiles of emissions associated with the model streamer along the central axis. The z coordinate is aligned with the direction of the electric field.

Densities of electrons and $N(^{2}D)$ are very similar. The density of $N_2(a^1\Pi_g)$ has a distribution exponentially decreasing from the streamer head to the streamer body because of the fast deactivation by radiative transition and quenching. The shape of the NO($A^2\Sigma^+$) density profile follows that of $N_2(A^3\Sigma_u^+)$ because it is produced by reaction 10 (Table 1) and the NO density is constant. For the emission intensity profiles, NO- γ emissions are approximately one order of magnitude weaker than N2 LBH emissions in the streamer body, but stronger than $1NN_2^+$. In the streamer head, N2 LBH emissions definitely dominate over NO- γ emissions. The relatively constant intensity ratio of NO- γ to N₂ LBH emissions in the streamer body is due to two factors: 1) Around the streamer body, there is a shell with the peak electric field about $\sim E_k$ (the conventional breakdown threshold field [see Liu and Pasko, 2005, Figure 2]). The N₂($a^1\Pi_g$) molecules excited in this shell continue to supply the N₂ LBH emissions after the N₂($a^{1}\Pi_{g}$) molecules produced by the passage of the streamer head are deactivated in a short time period (the lifetime of $N_2(a^1\Pi_g)$ is 14 µs at 70 km altitude [Liu and Pasko, 2005]); 2) The metastables $N_2(A^3\Sigma_u^+)$ in the streamer body, which are also produced by the passage of the streamer head, continuously excite NO ground state to NO($A^2\Sigma^+$) leading to NO- γ emissions. The total lifetime of N₂($A^{3}\Sigma_{\mu}^{+}$) at 70 km altitude accounting for quenching effects (reactions 12 and 13 in Table 1) is approximately 1 ms, which is longer than the 0.44 ms simulation time.

[12] Although the model streamer is very short, it is possible to estimate the relative strength of the total NO- γ emissions with respect to the total N₂ LBH emissions for a long streamer or even for an entire sprite event. Generally, it is expected that excitation of $N_2(A^3\Sigma_u^+)$ would be stronger than $N_2(a^1\Pi_g)$ because the excitation energy threshold for N₂($A^{3}\Sigma_{u}^{+}$) (6 eV) is lower than N₂($a^{1}\Pi_{g}$) (8.55 eV). However, according to Figure 2, the two excitation frequencies are very close to each other. As a matter of fact, the ratio of N₂($A^{3}\Sigma_{u}^{+}$) to N₂($a^{1}\Pi_{g}$) excitation frequencies varies from 2.4 to 0.6 for the electric field range of interest (> 10^6 N/N_0 V/m, where N is the air density at the altitude under consideration and N_0 is the density at ground level). Therefore, it can be assumed, to the first order approximation, that the production of both species by electron impact reactions is on the same level in sprites, i.e., for our analysis here we assume that equal numbers of $N_2(A^3\Sigma_u^+)$ and $N_2(a^1\Pi_g)$ molecules are produced in the streamer discharges. For N₂($A^{3}\Sigma_{u}^{+}$) species, the loss processes are associated with reactions 8, 10, 12 and 13 (Table 1). Reaction 8 will be ignored in the following analysis, because the density of $O({}^{3}P)$ is much lower than those of NO and O₂ at 70 km altitude. Using a density of 3.83 \times 10^{20} m⁻³ for O₂ and the density range 10^{13} - 10^{14} m⁻³ for NO at 70 km altitude, the $N_2(A^3\Sigma_u^+)$ deactivation frequencies associated with reactions 10, 12 and 13 are 0.001-0.01, 335 and 624 1/s, respectively. Therefore, only about 0.0001–0.001% of produced N₂($A^{3}\Sigma_{\mu}^{+}$) species transfers energy to NO ground state molecules to form NO($A^2\Sigma^+$). Because the quenching altitude of NO($A^2\Sigma^+$) is much lower than 70 km, essentially all NO($A^2\Sigma^+$) molecules are deactivated by emitting NO- γ emissions. The quenching altitude of $N_2(a^1\Pi_g)$ is about 77 km. At 70 km altitude, the ratio of loss rates by radiative transition and quenching is about 1/e. Therefore, at 70 km altitude about 25% of produced N₂($a^{1}\Pi_{o}$) species leads to N₂ LBH emissions. Hence, the total production of NO- γ emissions is



Figure 2. Excitation frequencies of $N_2(A^3\Sigma_u^+)$ [Aleksandrov et al., 1995] and $N_2(a^1\Pi_g)$ [Moss et al., 2006] as a function of reduced electric field. N is the air density at the altitude under consideration and N_0 is the density at ground level.

substantially weaker in comparison with N₂ LBH emissions for a streamer at 70 km altitude. Since the total intensity of N_2 LBH is more than 4 orders of magnitude greater than that of the NO- γ emissions, the NO- γ emissions from sprites are not observable for a wide bandwidth photometer such as the one used by the ISUAL instrument [Kuo et al., 2005]. This result is notably different from the laboratory observations [e.g., Teich, 1993; Tochikubo and Teich, 2000], and analysis of the reasons leading to this difference will be given at the end of this paper. We note that strong bands of NO- γ emissions are located in the wavelength range 240– 260 nm in which N2 LBH emissions are absent [Vallance-Jones, 1974, Tables 4.14 and 4.18]. It is therefore possible that a dedicated narrow bandwidth photometer with the wavelength passband of 240-260 nm would be able to detect sprite NO- γ emissions from space (see [Liu et al., 2006, Figure 3] regarding the atmospheric transmittance in this wavelength range).

[13] The same approach can be applied to analyze relative strength of NO- γ emissions to N₂ LBH emissions for elves and aurora.

[14] Assuming that the typical altitude of elves is 100 km [*Mende et al.*, 2005, and references therein], quenching processes are still the dominating loss mechanism of $N_2(A^3\Sigma_u^+)$ molecules. At this altitude, the number densities of O, NO and O₂ are 4×10^{17} , 2×10^{13} and 2.17×10^{18} 1/m³, respectively. The $N_2(A^3\Sigma_u^+)$ deactivation frequencies for reactions 8, 10, 12 and 13 are 2.79, 1.8×10^{-3} , 1.9 and 3.53 1/s, respectively. Therefore, only $\frac{1.8 \times 10^{-3}}{2.79 + 1.8 \times 10^{-3} + 1.9 + 3.53} \times 100\% = 0.02\%$ of $N_2(A^3\Sigma_u^+)$ molecules excite NO($X^2\Pi_r$) leading to the NO- γ emissions. However, this altitude is higher than the quenching altitude of N_2 LBH emissions and all of excited $N_2(a^1\Pi_g)$ molecules emit. We therefore can draw a conclusion that N_2 LBH emissions are also much stronger than NO- γ emissions for elves.

[15] For aurora, we pick the altitude where the ambient NO density profile is maximum $(3 \times 10^{13} \text{ 1/m}^3 \text{ at} \sim 120 \text{ km})$ [*Minschwaner et al.*, 2004]. At this altitude, the number densities of O and O₂ are 7.914 × 10¹⁶ and 4.395 × 10¹⁶ 1/m³, respectively. The N₂($A^3\Sigma_u^+$) deactivation frequencies for reactions 8, 10, 12 and 13 are 0.554, 2.6 × 10⁻³, 0.0385 and 0.0716 1/s, respectively. The loss of N₂($A^3\Sigma_u^+$) due to radiative transition (i.e., N₂ Vegard-Kaplan bands with Einstein coefficient $A_k = 0.5$ 1/s) is also important. About $\frac{2.6 \times 10^{-3}}{0.554 + 2.6 \times 10^{-3} + 0.0385 + 0.0716 + 0.5} \times 100\% = 0.22\%$ of

produced N₂($A^3\Sigma_u^+$) molecules excites NO($X^2\Pi_r$). However, it is known that NO- γ emissions have been observed from aurora [e.g., *Vallance-Jones*, 1974, p. 140]. The explanation for the observation is that ambient NO density can be enhanced in aurora and it can be as high as 10^{15} 1/m³ [*Viggiano and Hunton*, 1999]. If this density is used, about 7% of the total production of N₂($A^3\Sigma_u^+$) excites NO($X^2\Pi_r$). This value is on the same order as the ratio ($\frac{63}{383} \times 100\% =$ 16.45% [*Vallance-Jones*, 1974, Tables 4.14 and 4.18]) of the total intensity of NO- γ and N₂ LBH emissions for IBC3 aurora. It should be noted that the present analysis assumes that the production of N₂($A^3\Sigma_u^+$) and N₂($a^1\Pi_g$) is mainly by electron impact excitation.

[16] We conclude by noting that strong NO- γ emissions from streamer discharges are observed in laboratory experiments [e.g., *Teich*, 1993; *Tochikubo and Teich*, 2000; *Ono*

and Oda, 2005]. In those experiments, streamer discharges are generally operated in repetitive mode and the metastable molecules $N_2(A^3\Sigma_{\mu}^+)$ accumulate during a discharge period of several hundreds of ns (100 ns scales to 1.5 ms at 70 km altitude). Additionally, a large amount of background NO is commonly added to the gas medium, for example 200 ppm in the work by Teich [1993]. As part of this study, we simulated a positive streamer at ground pressure with a background NO density 200 ppm. The results (not shown here for the sake of brevity) indicate that NO- γ emissions are several orders of magnitude stronger than N₂ LBH emissions in the streamer body and slightly weaker in the streamer head for a positive streamer propagating up to 20 ns. The distribution of intensity of NO- γ emissions is almost constant from the streamer head to the body (similar to Figure 1b but with different magnitude). Therefore, under this kind of conditions, the overall intensity of NO- γ emissions is stronger than N₂ LBH emissions, in agreement with Teich [1993].

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